

# Impact of Rubbing and Waiting Technique of a Universal Adhesive System on Shear Bond Strength of Orthodontic Brackets: A Comparative Analysis

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#### Abstract

**Background:** One important factor in orthodontics to consider is the bracket's bonding strength to the teeth substrate. This study's objective is to see how the rubbing and waiting approach of a universal adhesive system affects the shear bond strength and adhesive remnant index (ARI) of orthodontic brackets.

**Materials and Methods:** This was an experimental study (in vitro). Fifty newly extracted human maxillary first premolars were divided into five groups of 10 teeth each. The universal adhesive was applied on the buccal surface with 20 seconds of rubbing followed by 0, 5, 10, 15, or 20 seconds of waiting protocol according to their group. The teeth were set into blocks of acrylic resin, and the shear bond strength was tested. Afterwards, ARI was assessed. Statistical analyses were performed using the one-way analysis of variance and Tukey HSD post hoc. In addition to the Kruskal-Wallis test.

**Results:** One-way ANOVA showed significant differences (p=0.000) between the groups. The maximum shear bond strength was for group 5 (18.93  $\pm$  2.82), with significant differences from group 1 (p= 0.000) and group 2 (p= 0.006). The minimum bond strength was for group 1 (11.09  $\pm$  1.50), with significant differences from all the other groups. ARI scores revealed no significant differences between the groups (p= 0.406).

**Conclusion:** By allowing for adequate solvent evaporation and monomer infiltration—both of which are essential for the clinical outcome of orthodontic treatment—extending the waiting period improves the shear bond strength. The ARI was not significantly impacted by waiting time.

**Keywords:** Orthodontics; Shear bond strength; Universal Adhesive.

Citation: Al-Fadhily ZM, et al. (2025) Impact of Rubbing and Waiting Technique of a Universal Adhesive System on Shear Bond Strength of Orthodontic Brackets: A Comparative Analysis. Dentistry 3000.1:a001 doi:10/5195/d3000.2025.812 Received: December 27, 2024 Accepted: February 9, 2025 Published: February 14, 2025 Copyright: ©2025 Al-Fadhily ZM, et al. This is an open access article licensed under a Creative Commons Attribution Work 4.0 United States License. Email: ZahraaM.AL-fadhily@uokufa.edu.iq

#### Introduction

Fixed appliances are among the most important orthodontic tools used to give patients more aesthetically beautiful and appealing smiles, which enhances their function and self-image [1]. In fixed orthodontic mechanics, orthodontic brackets are the most crucial passive components that transfer the force generated by the force elements to the teeth [2].

objective fixed One key of orthodontic therapy is creating a suitable bond between orthodontic brackets and tooth surfaces. The bond must be sufficiently robust to withstand the forces that arise during treatment and permit bracket take-off after therapy without causing harm to the tooth enamel [3]. Orthodontic bond strength depends on a variety of factors. regarding the materials,

details are provided about the enamel etching technique, bracket layout, and kind of adhesive agent; regarding the teeth, details are provided on the type of tooth and fluorosis; regarding the surroundings, details are provided about the moisture, saliva, and blood contamination [4-6].

Fixed appliance attachments should be firmly bonded to enamel surfaces to resist various events in the oral environments. Frequent debonding of orthodontic brackets leads to therapy interruption and lengthens treatment duration which causes discomfort for orthodontists and their patients [7]. Over the years, research has been done on direct bracket bonding to dental enamel. In an effort to enhance and achieve sufficient bond strength in orthodontics, evaluations of bonding systems and various enamel surface preparation techniques have been carried out [8-10].

Adhesive systems referred to as "universal" or "multimode" were introduced into the market. These materials exhibited less а aggressive pH level of 2 to 3 and accomplished adhesion through the use of functional monomers [11]. These substances may be applied to the substrate with or without prior etching [12,13]. Universal adhesives differ from standard one-step self-adhesives as they contain functional phosphate and/or carboxylate monomers. Several of these functional monomers can initiate chemical bonding with calcium in hydroxyapatites [14,15].

The universal adhesive was successfully employed for bonding orthodontic brackets, especially with etching mode [16-18].

This study aims to evaluate the impact of the rubbing and waiting technique of a universal adhesive system on the shear bond strength and the adhesive remnant index of orthodontic brackets.

### **Material and Methods**

Specimen Collection and Preparation

For this experimental study, 50 newly extracted human maxillary first premolars were collected from three orthodontic clinics. Collected from healthy 15–30-year-old donors their teeth needed to be extracted as a part of orthodontic treatment plans. Donors provided written agreements for the use of extracted data.

The teeth sample had to meet certain requirements to be included: no apparent decalcification, cavities, cracks, hypoplastic areas, or restorations (the teeth were checked using a light curing unit) [19].

The teeth were cleared of debris and preserved in a closed container with normal saline at room temperature (220±3). The normal saline was regularly replaced to prevent bacterial development until the bonding procedure.

Before the bonding procedure, all teeth' buccal surfaces were cleaned and given a 20-second polish with non-fluoridated pumice. Before mounting, teeth were irrigated with a water spray for ten seconds and then dried with three air syringes for ten seconds.

# Mounting Technique

To improve the teeth's retention inside the acrylic blocks, a big fissure bur was utilized to create retentive grooves that serrated every tooth. Afterward, the teeth were positioned vertically in a selfcure acrylic mounting, exposing the crown and directing the buccal surface to run parallel to the analyzing survevor rod. This resulted in the application of force at a 90<sup>0</sup> angle to the bracket-tooth contact.

Specimen Grouping and Bonding Technique

The buccal surfaces of the mounted specimens were etched for 20 seconds using phosphoric acid gel (37%) (Ortho-Technology; FI/USA) followed by water rinsing and drying [20].

Following that, the teeth were divided into five groups randomly according to the bonding technique with ten teeth per group (Figure 1).

Group 1: The adhesive (3M<sup>™</sup> Scotchbond<sup>™</sup> Universal Adhesive) was applied on the buccal surface with 20 seconds of rubbing, air drying for five seconds, and 10 seconds of curing utilizing LED curing light Valo Ortho (Ultradent, UT). This protocol was according to the manufacturer's instructions.

Group 2: The adhesive was put on the buccal surface with 20 seconds of rubbing, waiting for 5 seconds, 5

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seconds of air drying, and 10 seconds of curing.

Group 3: The adhesive was put on the buccal surface with 20 seconds of rubbing, waiting for 10 seconds, 5 seconds of air drying, and 10 seconds of curing.

Group 4: The adhesive was put on the buccal surface with 20 seconds of rubbing, waiting for 15 seconds, 5 seconds of air drying, and 10 seconds of curing.

Group 5: The adhesive was put on the buccal surface with 20 seconds of rubbing, waiting for 20 seconds, 5 seconds of air drying, and 10 seconds of curing.

Following that, 50 Maxillary first premolar stainless-steel brackets (Dentarum company, Ispringen, Germany) were attached to the teeth using orthodontic adhesive paste (Transbond XT, 3M). The then bracket was carefully positioned on the tooth using the least force possible and pushed ensure firmly to precise positioning. Next, use a sharp scalar to remove any leftover adhesive paste that may have flown out from behind the bracket.

Maxillary First Premolars (50) divided into 5 Groups
Group 1:20 seconds of rubbing and 10 seconds of curing
Group 2:20 seconds of rubbing, waiting for 5 seconds, 5 seconds of air drying, and 10 seconds of curing
Group 3: 20 seconds of rubbing, waiting for 10 seconds, 5 seconds of air drying, and 10 seconds of curing.
Group 4: 20 seconds of rubbing, waiting for 15 seconds, 5 seconds of air drying, and 10 seconds of curing.
Group 5: 20 seconds of rubbing, waiting for 20 seconds, 5 seconds of air drying, and 10 seconds of curing.

Figure 1. A diagram displaying the specimen grouping and sample size.

Shear Bond Strength Experiment

A universal testing machine (model 4411; Instron, USA) was employed to measure the force required to debond the brackets (Figure 2). The preparation of the specimens involved positioning them in acrylic resin with brackets orthogonal to shear blade. All five the experimental groups underwent a shear bond strength test, wherein the active shear blade tip was positioned on the upper part of the bracket base (Figure 3). The debonding force was applied to the bracket-tooth interface using a blade operating at a crosshead velocity of 0.5 mm/minute [21].

The measurements were obtained in kilogram-force (kg), converted to Newton (N), and the shear bond strength was estimated by dividing the amount of force by the bracket surface area. The bracket base's average surface area was 14 mm2, which allowed for the measurement of the shear bond strength in MPa.



Figure 2. Universal testing machine.



Figure 3. Shear bond experiment.

# Adhesive Remnant Index Evaluation

After the debonding process, each tooth surface and the bracket base were evaluated by a stereomicroscope (Olympus, Tokyo, Japan) with an X20 magnification for evaluation of the adhesive remnant index (ARI) (Figure 4), based on the ARI scores (Table 1) [17].

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Figure 4. Microscopic evaluation of the Adhesive Remnant Index.

Table 1: Scores for the Adhesive<br/>Remnant Index.Score 0: Indicates that no adhesive<br/>remains attached to the enamel.Score 1: Less than ½ of the adhesive<br/>remains attached to the enamel.Score 2: More than ½ of the adhesive<br/>remains attached to the enamel.Score 3: The whole adhesive remains<br/>attached to the enamel.

#### **Statistical Analyses**

The statistical analyses were conducted utilizing SPSS version 26. Mean, minimum, maximum, standard deviation, and standard error were employed to evaluate the data. Shapiro-Wilk and Levene's tests, respectively, were used to screen the data for homogeneity and normal distribution. Inferential statistics utilize a 95% confidence interval. A p-value was deemed statistically significant if it was less than 0.05. As the data of shear bond revealed

a normal distribution according to the Shapiro-Wilk test, One-way ANOVA and Tukey HSD post hoc multi-comparison test were used to test if there were significant differences between groups. Since the data was not continuous and did not follow a normal distribution, the ARI scores were examined by applying the Kruskal-Wallis test.

# Results

The results of descriptive statistics included Min, Max, M, SD, SE, and Shapiro-Wilk test values of shear bond strength for all studied groups demonstrated in Table (2). The data were considered to satisfy the normality criteria within the 5% significance level based on the Shapiro-Wilk test, which found that the P value for all groups was greater than the significance level of 0.05. Levene's test indicated that the data were homogeneous. One-way ANOVA revealed significant differences (p=0.000) between the groups (Table 3).

The highest shear bond strength was for group 5 (18.93  $\pm$  2.82), with significant differences from group 1 (p= 0.000) and group 2 (p= 0.006). The lowest bond strength was for group 1 (11.09  $\pm$  1.50), with significant differences from all the other groups. Group 2 differs considerably from all other groups, but not significantly from Group 3 (p= 0.195) (Table 4). The ARI score

variation for each of the five groups is shown in Table (5). The statistical examination of the ARI scores with the Kruskal-Wallis test showed no significant differences between the groups (p= 0.406). Vol 13 No 1 (2025) DOI 10.5195/d3000.2025.812

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<b>Table 2.</b> Descriptive data of shear bond strength (MPa) for all groups.							
Groups	N	Min	Max	Mean	SD	SE	Shapiro- Wilk test (P-value)
Group 1	10	8.99	13.30	11.09	1.50	0.47	0.487
Group 2	10	11.99	17.80	15.16	1.96	0.62	0.426
Group 3	10	14.00	22.00	17.44	2.33	0.73	0.891
Group 4	10	15.30	24.00	18.75	2.67	0.84	0.491
Group 5	10	14.56	23.80	18.93	2.82	0.89	0.718

Table 3. One-way ANOVA test of she	ar bond strength of a	ll groups.			
	Sum of Squares	df	Mean Square	F	p-\

	Sum of Squares	dt	Mean Square	F	p-value
Between Groups	426.836	4	106.709	19.959	0.000
Within Groups	240.593	45	5.347		
Total	667.428	49			



**Table 4.** Tukey HSD post hoc multiple comparison test of shear bond strength of all groups.

Groups		Mean	SE	Sig.	95% Confidence Interval		
		Difference			Lower bound	Upper bound	
	Group 2	-4.067*	1.03	0.003	-7.00	-1.12	
Group 1	Group 3	-6.351*	1.03	0.000	-9.28	-3.41	
	Group 4	-7.662*	1.03	0.000	-10.60	-4.72	
	Group 5	-7.841*	1.03	0.000	-10.77	-4.90	
	Group 1	4.067*	1.03	0.003	1.12	7.00	
Group 2	Group 3	-2.284	1.03	0.195	-5.22	0.65	
	Group 4	-3.595*	1.03	0.010	-6.53	-0.65	
	Group 5	-3.774*	1.03	0.006	-6.71	-0.83	
	Group 1	6.351*	1.03	0.000	3.41	9.28	
Group 3	Group 2	2.284	1.03	0.195	-0.65	5.22	
	Group 4	-1.311	1.03	0.712	-4.24	1.62	
	Group 5	-1.490	1.03	0.605	-4.42	1.44	
	Group 1	7.662*	1.03	0.000	4.72	10.60	
Group 4	Group 2	3.595*	1.03	0.010	0.65	6.53	
	Group 3	1.311	1.03	0.712	-1.62	4.24	
	Group 5	-0.179	1.03	1.000	-3.11	2.75	
	Group 1	7.841*	1.03	0.000	4.90	10.77	
Group 5	Group 2	3.774*	1.03	0.006	0.83	6.71	
	Group 3	1.490	1.03	0.605	-1.44	4.42	
	Group 4	0.179	1.03	1.000	-2.75	3.11	

Table 5	5. Scores for e	ach group of	the adhesive	remnant inde	x (ARI).
Groups	Score 0	Score 1	Score 2	Score 3	P-value
Group 1	2	4	2	2	0.406
Group 2	2	5	3	0	
Group 3	1	4	5	0	
Group 4	0	6	3	1	
Group 5	1	6	3	0	

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#### Discussion

One advantage of the self-etching adhesive system is that it requires fewer stages and lowers the possibility of mistakes during the adhesive method [22]. The methacrylate phosphoric acid esters found in these adhesive systems are typically produced from phosphoric acid and demineralize the surface of the tooth by removing calcium ions [23,24]. One important factor to consider while evaluating an adhesive's efficacy is its bond strength. The adhesive strength between the two substrates is assessed using the shear bond strength test. The load that a material can withstand in a direction parallel to its face is measured by the shear bond strength test [25]. the adhesion strength can be affected by: The chemical makeup of the adhesive, light-curing apparatus, acid content, and variations in the experimental methodology [26-28]. though Even two-step

complete etch bonding techniques can yield high initial resin-dentin bond strength values, they still undergo substantial decline after in vitro aging [29].

Because of the hydrophilic nature of the polymer inside the hybrid layer that promotes water sorption and increases permeability, it is vulnerable to hydrolytic deterioration [30,31]. With time, the adhesive-formed hybrid layer could deteriorate because of an increase in water absorption and a loss of collagen fiber cross bands, weakening the bond between the surfaces. This time frame could range from six months to three to five years [26,32]. In addition, research revealed that insufficient resin monomer infiltration resulted in a significant number of exposed collagen fibrils at the hybrid layer [29]. dentin-restorative The material's connection is weakened when the collagen fibrils are targeted by cysteine cathepsins and matrix metalloproteinases (MMPs) [33]. Current studies

established and highlighted the role of MMPs and cysteine cathepsins on the degradation of dentin bonding after using both total-etch bonding systems and self-etch [34-36].

In order to overcome the complex nature of dentin, researchers have number of proposed а modifications to the recommended application techniques that should make the application easier to apply and maximize infiltration. Some of these modifications include longer light exposure times for bonding resins [37], multiple adhesive coatings with delaved polymerization, adhesive rubbing, and increased application times for primers and bonding resins [38, 39]. It doesn't seem to be entirely that self-etch true bonding solutions can penetrate along all demineralized dentin [34,35]. If the bonding resin is applied for a longer period of time, there will be a longer period of uninterrupted waiting before light curing. It was

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anticipated that this amount of waiting time would help with the evaporation of leftover solvents and unbound water [37].

Additionally, this approach change would facilitate the creation of a strongly cross-linked polymer and aid in the decrease of nanopores. As a consequence, the resindentin hybrid layer that is produced would be of higher quality, stronger bonds, and more resistant to deterioration [36]. The active application of the substance by rubbing has already been discussed concerning the application of resin to dentin and enamel. The main objectives are to enhance the adhesive-substrate contact surface and promote solvent evaporation [27,40].

The result showed highest shear bond strength was for group 5 in which the waiting time extended to 20 seconds with significant differences from other groups with less waiting time. The lowest bond strength was for group 1 which included waiting for 5 seconds only. So, the result declared that increased waiting time of bonding before curing can increase the bonding strength. Waiting time permits for solvent evaporation and better infiltration of monomer. different solvents, like: ethanol, water, or acetone, are included in the universal adhesive system to aid in the adhesive's transfer to the enamel. To prevent residual solvent from weakening the adhesive layer and interfering with the polymerization process, proper solvent evaporation is essential.

If there is not enough waiting time before light curing, the residual solvent may get trapped in the adhesive layer, which can cause voids and weaker bonding. Depth of monomer penetration is more likely when the adhesive has enough time to interact with the enamel. Additionally, it provides time for the adhesive to level out and create a consistent layer, which is essential for guaranteeing proper mechanical stability and polymerization.

It has been shown in several earlier investigations that the quick curing of bonding resin, without a waiting interval for the resin monomers to permeate into the etched dentin formed by total-etch bonding methods, is only partially effective. Similarly, it doesn't seem to be entirely true that Self-etch bonding techniques can penetrate along all demineralized dentin [34]. Marchesi et al. (2014) found that a short waiting period between the application of adhesive and curing could weaken the binding. For sufficient solvent evaporation and adhesive contact with the substrate, the authors advised

waiting for at least 10 seconds [35]. Studies reveal that using a rubbing motion when applying adhesive the orthodontic enhances brackets' shear bond strength (SBS). Rubbing improves micromechanical retention by facilitating the entry of sticky into the monomers enamel. Furthermore, improved contact with the smear layer is achieved through active application, especially in self-etch systems, which is essential for forging a robust adhesive interface [13]. In contrast to passive applications, adhesive active system applications greatly increased binding strengths, according to a study by Papadogiannis et al. (2019). The results of the investigation demonstrated а correlation between improved adhesive penetration and higher SBS, which strengthened the binding between the enamel surface and orthodontic brackets [41]. According to a study by Hanabusa et al. (2012), the most dependable orthodontic bonding procedure produced the highest SBS values when a universal adhesive system was applied using both active application and a waiting period prior to curing [13].

Regarding the adhesive remnant system (ARI), the result cleared that increasing the waiting time had no effect on the ARI scores and

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differ The did not groups significantly from one another. This result was approved by the guantities of adhesive left on the surfaces did enamel not significantly increase with longer application times or more agitation of the self-etching primer, according to Protásio et al. (2016). These findings, along with those of a different study, imply that longer application times and more agitation should not result in a higher risk of enamel fracture or require more time for tooth cleanup following deboning [24,42].

# Conclusion

Increasing the waiting time enhances the shear bond strength by allowing for proper solvent evaporation and monomer infiltration, which is critical for clinical success in orthodontic treatment. Waiting time did not produce a significant effect on the ARI.

# Practical Implications in Orthodontics

Combining rubbing and waiting techniques can be a standard orthodontics, practice in particularly when using universal adhesive systems. Orthodontic brackets need withstand to substantial forces, and a strong, durable bond is crucial for treatment efficacy and the longevity of the bond.

#### Availability of Data

This article contains the data used to support the study's conclusions.

#### **Conflicts of Interest**

According to the authors, there are no conflicts of interest.

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